clude any physically relevant Shubnikov groups. The 28 non-crystallographic Shubnikov groups discussed above are listed in Table 2.

References

BERTAUT, E. F. (1968). *Acta Cryst.* A 24, 217. KOPSTIK, V. A. (1966). *Shubnikov Groups*, p. 29. Moscow Univ. Press.

- MURNAGHAN, F. D. (1938). The Theory of Group Representations, chap. 10. London: Constable.
- SAMSON, S. (1968). In Structural Chemistry and Molecular Biology. Ed. A. RICH & N. DAVIDSON. p. 687. San Francisco and London: Freeman.
- TAVGER, B. A. & ZAITSEV, V. M. (1956). Zh. éksp. teor. Fiz.
 30, 564. Transl. R. E. PEIERLS, Soviet Phys. JETP, 3, 430.
 ZAMORZAEV, A. M. (1953). Generalization of the Space Groups. Dissertation: Leningrad Univ.

Acta Cryst. (1969). A 25, 459

High-Resolution Monochromatization of Neutrons by Multiple Bragg Reflection in Hexagonal Close-Packed Crystals*

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The conditions which give high-resolution monochromatization of neutrons by multiple Bragg reflection in hexagonal close-packed crystals have been studied theoretically. The role of reflections forbidden by special atomic positions has been clarified, and, contrary to a previous conclusion, it is found that these 'forbidden' reflections may be useful for high-resolution purposes. Detailed calculations of the necessary crystal orientations have been carried out for beryllium, magnesium and zinc. On the basis of somewhat arbitrary criteria the calculations show that these crystals can reflect a total of approximately 200 wavelengths in the range 0.7 to 4.9 Å (neutron energy 0.003 to 0.17 eV). Several orientations have been found at which *mosaic* crystals (with a sacrifice of angular resolution) should produce beams that are both intense and highly monochromatic.

Introduction

In a previous paper (hereinafter referred to as part I) it was pointed out that under suitable conditions in certain types of perfect crystals the phenomenon of multiple Bragg reflection (MBR) can provide highly monochromatic and highly collimated (semiparallel) beams of neutrons and X-rays at fixed wavelengths (Kottwitz, 1968a). The effect depends on the simulation of a forbidden (primary) reflection by the cooperative action of two allowed reflections (secondary and tertiary); this is the 'Umweganregung' phenomenon (Renninger, 1937). For each such simulation the reflected wavelength has a second-order extremum at a definite orientation (referred to in part I as the 'operating point') of the incident beam relative to the crystal. At such operating points even quite coarse external collimation can produce extremely high wavelength and angular resolution provided there is no 'interference', that is, provided no other simulation is close enough to make an impure contribution. Exploratory intensity measurements have been reported (Kottwitz, 1968b).

The main purpose of this paper is to present and discuss calculations of interference-free operating

points for three hexagonal close-packed crystals: beryllium, magnesium and zinc. In contrast with the cubic diamond structure, for which the operating points are independent of lattice parameters, the h.c.p. structure permits various c/a ratios and thus requires separate calculations for each crystal. These particular crystals were chosen because they are suitable monochromators for neutrons (Bacon, 1962).

The general equations, nomenclature and conventions that provide the basis for this paper are identical with those in part I, and will be at most briefly described here. Attention will be concentrated on characteristics particularly relevant to the h.c.p. case.

Forbidden reflections in h.c.p. crystals

The h.c.p. structure (space group $P6_3/mmc$) has some reflections that are strictly forbidden by space-group symmetry; they are given by $hh\overline{2h}l$ with l odd. Calculations of operating points have been done for 0001, 0003, 0005, 11 $\overline{2}1$, 11 $\overline{2}3$ and 22 $\overline{4}1$. These are the preeminent candidates for use in high-resolution monochromatization by MBR.

The h.c.p. structure also has reflections that are only approximately 'forbidden' by special atomic positions; these are given by *hkil* with *l* odd and $h-k=3n\neq 0$. They are represented in our calculations by $30\overline{3}1$, $30\overline{3}3$

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or

and 4151. Furthermore at sufficiently low temperatures the operating points for these reflections probably are useful for high-resolution monochromatization of neutrons.

Since the last remark is contrary to the conclusion reached in § 5 of part I, it is necessary to clarify this point. It is customary to specify special positions for spherical scattering centers (International Tables for X-ray Crystallography, 1952). However, the condition of full spherical symmetry, while sufficient, is not necessary to nullify a reflection. Thus some of the special positions may be effective for less symmetrical scattering centers. This is so for the h.c.p. structure, since the elementary structure-factor calculation (Guinier, 1952) remains valid for all the 'forbidden' reflections when spheres are replaced by ellipsoids of revolution having axes parallel to the c axis. Now the nuclear 'cloud' in h.c.p. crystals has just this shape in the harmonic approximation, that is, at sufficiently low temperatures. The approximately 'forbidden' reflections may then become truly forbidden, thus making possible high-resolution monochromatization of neutrons by MBR. The electronic distribution, being more sensitive to the crystal structure, is likely to have an irregular shape; thus high-resolution monochromatization of X-rays by these approximately 'forbidden' reflections would not be possible. Similarly, at high temperatures the nuclear 'cloud', responding to the crystal structure, is not ellipsoidal, and high-resolution monochromatization of neutrons is no longer possible. In fact such weak 'forbidden' reflections are particularly useful in studying the spatial aspects of anharmonicity by neutron diffraction (Dawson & Willis, 1967).

Calculations for h. c. p. crystals

The calculations for Be, Mg and Zn were done on a UNIVAC 1108 computer with the FORTRAN V program *DBSKH2*. The program has essentially the same basic structure as that used in Part I for diamondstructure crystals. The only significant change was to add the calculation of θ_2 and θ_3 , the Bragg angles (at the operating point) of the secondary and tertiary reflections, respectively. In the notation of Part I these quantities are given by

$$\tan \theta_2 = R/|\mathbf{K}_1 \cdot \mathbf{K}_3| \tag{1a}$$

$$\tan \theta_3 = R/|\mathbf{K}_1 \cdot \mathbf{K}_2| \ . \tag{1b}$$

They are related to the primary Bragg angle θ_p by

$$\theta_p = |\theta_2 - \theta_3| , \qquad (2a)$$

 $\theta_p = \text{smaller of } (\theta_2 + \theta_3) \text{ and } (180^\circ - \theta_2 - \theta_3), (2b)$

the former equation applying when the senses of the secondary and tertiary reflections are opposite (parallel case) and the latter when they are similar (antiparallel case). The Bragg angles θ_2 and θ_3 provide useful semiquantitative information about the resolution of a simulated reflection.

The physical constants used in the calculations had the 1963 IUPAP values as follows: Planck's constant = 6.6256×10^{-34} J-sec, neutron mass = 1.67482×10^{-27} kg and electron charge = -1.60210×10^{-19} C (*Physics Today*, 1964). Lattice parameters were taken from the recent compilation by Pearson (1967); they are shown in Table 1. It is interesting to note that these values of c/a accidentally cover the range of physical interest fairly well; Be has the smallest observed ratio, Zn has a value within 2% of the largest observed (Cd), while Mg has a value within 1% of that for close packing of spheres (1.6330).

Table 1. Lattice parameters used in calculations

Crystal	a (Å)	c (Å)	c/a
Be	2.286	3.584	1.568
Mg	3.2094	5.2105	1.6235
Zn	2.6649	4.9468	1.8563

A summary of some of the important input parameters and results for each simulated reflection is given in Table 2. The parameters α_{\min} , α_{\max} , θ_{\min} , and θ_{\max} , which specify the (α, θ) -rectangle to be scanned for operating points, were chosen as follows. The largest range of α ever required is 180°; however, if \mathbf{K}_1 is parallel to a symmetry axis or a symmetry plane, the necessary range of α may be reduced considerably.

Table 2. Summary of input parameters and results for simulated h.c.p. reflections

Column 2 gives the direct lattice vector that is perpendicular to the scattering plane when $\alpha = 0$. Columns 3-6 give the input parameters defining the (α, θ) -rectangle investigated. Columns 7-9 give the number of operating points (including coincident ones) found in the rectangle. Columns 10-12 give the number of operating points (excluding coincidences) found to be free of interference.

Simulated	Reference	$ heta_{ ext{min}}$	θ_{\max}	α_{\min}	α_{max}	I	Points tested			Points accepted			
reflection	vector Z	(°)	(°)	(°)	(°)	Be	Mg	Zn	Be	Mg	Zn		
0001	[010]	6	90	0	30	56	52	44	4	4	5		
0003	[010]	14	90	0	30	112	104	86	8	5	7		
0005	[010]	23	90	0	30	114	106	86	9	8	11		
1121	[T10]	21	90	0	90	107	112	130	13	15	11		
1123	[T10]	28	90	0	90	104	101	104	11	11	12		
2241	[T10]	40	90	0	90	109	114	128	3	6	4		
3031	[010]	35	90	0	90	117	120	137	7	9	7		
3033	[010]	40	90	0	90	109	112	118	7	11	10		
4131	[T 40]	61	90	- 90	90	117	119	134	1	0	0		

The value of θ_{\min} is chosen empirically by a sequence of calculations, which indicate that there are very few, if any, interference-free operating points at smaller values of θ . The results in columns 7-12 of Table 2 show that only about 10% of the tested operating points are free of interference.

Detailed results on the interference-free operating points are given in Tables 3, 4, and 5. In all of these interference calculations the size of the diamondshaped window was specified by: $\Delta \theta = 1^{\circ}$, and $\Delta \beta = 2^{\circ}$. The arbitrary criteria for accepting an operating point as interference-free were the same as in part I: θ_{-} and $\theta_+ \ge 0.25^\circ$, and β_- and $\beta_+ \ge 1^\circ$. A detailed definition of these quantities is shown in Fig. 5 of part I.

Discussion of results

In Tables 3, 4, and 5 we note two striking features previously seen in the case of germanium, that is, a large range of values of θ_p and the frequent occurrence of coincident operating points. In addition, however, there is a new characteristic not present in the case of germanium. For the 0001, 0003 and 0005 reflections of each crystal there is at least one simulation which

Table 3. Operating points free of interference for 9 simulated beryllium reflections

The Bragg angle θ_p and azimuthal angle α in columns 3 and 4 specify the high-resolution orientation relative to the simulated reflection; λ and E in columns 5 and 6 give the corresponding wavelength and energy. θ_- , θ_+ , β_- and β_+ in columns 7-10 give the size of the interference free region. In column 7 or 8, *** indicates that θ_- or $\theta_+ \ge 1^\circ$; and in column 9 or 10, *** indicates that β_- or $\beta_+ \ge 2^\circ$. Columns 11 and 12 give the values of the secondary and tertiary Bragg angles.

REFLECTION PRIMARY	N INDICES (HKIL) SECONDARY	THETA (DEG)	ALPHA (DEG)	LAMBDA (Å)	ENERGY (Ev)	THETA- (DEG)	THETA+	BETA- (DEG)	BETA+ (DEG)	THETA2 (DEG)	THETA3 (DEG)
0 0 0 1 0 0 0 1 0 0 0 1 0 0 1	1 0-1 0; 1 0-1 1 1 0-1-1; 1 0-1 2 2 0-2-1; 2 0-2 2 2 0-2-2; 2 0-2 3	28.91543 18.93405 13.47578 10.72875	.00000 .00000 .00000 .00000	3,46586 2,32587 1,67039 1,33439	•006810 •015121 •029317 •045940	*** *** *** •44	*** *** •31	*** *** ***	*** *** ***	61.08, 90.00 42.15, 61.08 61.08, 74.56 50.36, 61.98	90.00, 61.08 61.08, 42.15 74.56, 61.08 61.08, 50.36
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	76.76492 58.89120 44.35508 36.73372 34.45634 32.40983 23.56699 23.30220	.00000 .00000 .00000 19.10661 .00000 19.10661 .00000	2,32587 2,04572 1,67039 1,42905 1,35183 1,28062 ,95531 ,94517	•015121 •019547 •029317 •040056 •044763 •049880 •089635 •091567	*** *** *** •31 •80	*** *** •73 •55 •38 •74	*** *** *** *** *** 1 • 2	*** *** 1 • 3 *** 1 • 6 1 • 2	42.15, 61.08 31.11, 90.00 61.08, 74.56 24.35, 61.08 67.34, 78.21 42.15, 74.56 43.77, 67.34 74.56, 82.14	61.08, 42.15 90.00, 31.11 74.56, 61.08 61.08, 24.35 78.21, 67.34 74.56, 42.15 67.34, 43.77 82.14, 74.56
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	85.43541 73.25931 70.09626 68.55962 63.28913 54.72412 51.65883 40.21062 28.91543	.00000 .00000 .00000 .00000 19.10661 19.10661 13.89788 .00000	1.42905 1.37284 1.34797 1.33439 1.28062 1.17036 1.12442 .92553 .69317	.040056 .043403 .045020 .045940 .049880 .059720 .064701 .095495 .170248	*** *** *** *** *** •72 •33	*** *** *** *** * * * * * * * * * * *	*** *** *** *** *** 1 • 1 1 • 1	*** *** *** *** *** 1 • 1 1 • 1	61.08, 24.35 42.15, 31.11 19.90, 90.00 50.36, 61.08 42.15, 74.56 57.94, 67.34 50.13, 78.21 58.50, 81.29 61.08, 90.00	24.35, 61.08 31.11, 42.15 90.00, 19.90 61.08, 50.36 74.56, 42.15 67.34, 57.94 78.21, 50.13 81.29, 58.50 90.00, 61.08
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$1 0-1-1 \\ 2 0-2 0, 0 1-1 1 \\ 1 0-1-2 \\ 2 0-2-1 \\ 2 0-2-2 \\ 1 0-1-3 \\ 1-1 0-2 \\ 2 0-2-3 \\ 2 1-3-1 \\ 2 0-2-3 \\ 2 1-3-1 \\ 4-1-3-1 \\ 2 1-3-5 \\ 4-3-1-2 \\ 1 0-2 \\ 1 0 0 0 0 \\ 1 0 0 0 0 \\ 1 0 0 0 0 \\ 1 0 0 0 0 \\ 1 0 0 0 0 \\ 1 0 0 0 0 0 \\ 1 0 0 0 0 0 0 0 0 0 $	86.28675 64.04487 62.52236 58.61374 50.20001 47.66199 42.99696 42.26796 39.44392 37.95540 29.41242 28.33836 27.01827	32.34988 62.24368 20.80858 43.53389 32.34988 15.18720 43.53389 25.40981 20.80858 72.47598 62.24368 8.31590 69.39853	2.17335 1.95H25 1.95224 1.65924 1.67326 1.48526 1.48526 1.38569 1.33953 1.06956 1.03381 .98938	.017318 .021332 .021910 .023664 .029217 .031562 .037082 .038121 .042726 .045589 .071507 .076538 .083568	*** *** *** *** •72 •88 •65 52 •52 •26 •29	*** *** *** *** *** *** *** *** *** **	*** *** 1 • 1 *** *** 1 • 5 1 • 1 1 • 0	*** 1 • 3 *** *** *** *** *** *** 1 • 9 1 • 3 1 • 0 1 • 5	38.83 81.55, 34.40 46.65 76.98 74.92 51.90 33.98 73.93 70.83 66.12 86.99 70.44	54,48 34,40, 29.64 70.83 44,40 54,88 80,44 76,98 63,80 31,38 75,93 50,76 64,67 82,54
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	82.75818 81.41275 75.02886 74.92738 60.77750 56.70802 55.26342 54.40292 48.24992 43.83436 38.58885	51,40119 68,24168 39,86709 26,61511 15,55639 68,24168 78,71409 59,09139 76,53262 68,24168 17,38966	1.63859 1.63325 1.59570 1.59494 1.44155 1.38069 1.35739 1.34310 1.23231 1.14398 1.03025	.030466 .030666 .032126 .032157 .039364 .042911 .044397 .045346 .053867 .062507 .077068	*** *** •49 •29 *** *** •88 •76 *** ***	*** *** *** *** *** *** *** *** *** **	*** *** *** 1.0 *** 1.5 2.0	*** *** 1.3 *** *** 1.6 *** 1.5	59.17 70.47 53.71, 51.26 27.40 74.42 70.47 79.39 63.83 55.43 51.34 78.88	38.07 28.11 51.26, 23.77 77.67 44.80 52.82 45.34 61.77 76.32 84.82 62.53
2 2-4 1 2 2-4 1 2 2-4 1	2 0-2-4 3-1-2-4 1 0-1-5	66.19150 57.40269 54.95053	22.16506 39.17185 9.90356	1.03268 .95094 .92405	•076706 •090460 •095801	•51 •69 •26	.72 .65 .88	*** 1.1 1.3	1.7 1.3 1.1	51.01 55.88 43.28	62.79 66.72 81.77
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 3-1-2-1 \\ 1-1 & 0-1 \\ 1 & 0-1-3 \\ 2 & 0-2-4 \\ 1 & 0-1-6 \\ 4 & 0-4-2 \\ 4 & 0-4-3 \end{array}$	88.43138 88.07111 73.25931 62.05426 44.13969 43.45155 43.14562	41.00830 53.80190 .00000 .00000 .00000 .00000 .00000	1.29752 1.29727 1.24299 1.14665 .90394 .89269 .88765	.048589 .048608 .052945 .062216 .100111 .102651 .103821	•41 *** *** •69 •31 •38	*** •32 •92 *** •55 •69 •31	1.4 1.3 1.7 1.4 1.0 1.2 1.0	1.8 1.4 1.7 1.4 1.0 1.2 1.0	62.34 21.98 37.42 59.66 52.22 69.32 76.08	29.23 69.95 69.32 58.28 83.64 25.87 32.94
3 0-3 3 3 0-3 3 3 0-3 3 3 0-3 3 3 0-3 3 3 0-3 3 3 0-3 3	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	86.38150 81.63258 70.18661 69.65502 62.21760 54.65660 48.66059	74.40243 60.82456 74.40243 35.61920 .00000 .00000 .00000	1.15298 1.14299 1.08690 1.08322 1.02211 .94237 .86740	.061534 .062615 .069245 .069716 .078301 .092113 .108724	•28 *** •86 •93 *** •28	*** •79 *** •50 *** *** •62	*** *** 1.2 1.1 1.1 1.0 1.6	1.0 1.2 *** 1.2 1.1 1.0 1.6	51.91 49.80 18.28 86.29 29.98 80.99 48.33	41.71 48.57 51.91 24.06 87.81 44.36 83.01
4 1-5 1	2 1-3-6	64.95517	-5,98718	.77716	•135437	•42	***	1.6	***	56.34	58,70

has θ_2 (or θ_3)=90°. Thus one of the relevant pair of cooperating reflections is a complete back-reflection.*

* In order to show that such an intermediate back-reflection is impossible for diamond-structure simulations, we first note from equation (1b) that $\theta_3 = 90^\circ$, only if

$$\mathbf{K}_1 \cdot \mathbf{K}_2 \equiv h_1 h_2 + k_1 k_2 + l_1 l_2 = 0$$
.

However, the indices are restricted as follows: (1) h_1, k_1 , and l_1 are each even, and $h_1 + k_1 + l_1 = 4n + 2$; and (2) h_2, k_2 , and l_2 are each odd. These restrictions readily give $K_1 \cdot K_2 = 4m + 2 \neq 0$. The proof for θ_2 is analogous.

Now it is a well-known consequence of Bragg's equation that $d\lambda/d\theta = 0$ at $\theta = 90^{\circ}$. This means that the extremely good wavelength resolution characteristic of a perfect crystal can be obtained even with a mosaic crystal having a small but finite angular spread. Thus highly monochromatic beams of relatively high intensity can be produced. However, it must be noted that such a result is obtained by sacrificing angular resolution.

The foregoing discussion would adequately account for a simple operating point, that is, one corresponding

Table	. / .	Am an atting		funn	f ind and	Comoso a d	C (a + a d		und antiour
таріе	: 4. (Cheraling	r nounus	iree a	n interi	erence i	or	$\gamma \gamma mu$	aiea	magnesium	renections
		0 p 0	, , , , , , , , , , , , , , , , , , , ,	J	<i>y</i>	•. ••• J	.				

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REFLECTION INDICES (HKIL) PRIMARY SECONDARY	THETA (DEG)	ALPHA (DEG)	LAMBDA (Å)	ENERGY (EV)	THETA⊶ (DEG)	THETA+ (DEG)	BETA- (DEG)	BETA+ (DEG)	THE TA2 (DEG)	THETA3 (DEG)
0 0 0 1 1 0-1 0, 1 0-1 0 0 0 1 1 0-1-1, 1 0-1	1 28.07667 2 18.77597	.00000	4.90467 3.35419	•003401 •007271	***	*** ***	***	*** ***	61.92, 90.00 43.15, 61.92	90.00, 61.92 61.92, 43.15
0 0 0 1 2 0-2-1, 2 0-2 0 0 0 1 1 0-1-2, 1 0-1	2 13.14274 3 11.14651	.00000	2.36950 2.01457	•014570 •020156	*** •36	***	*** ***	*** ***	61.92, 75.07 32.00, 43.15	75.07, 61.92 43.15, 32.00
0 0 0 3 1 0-1 1, 1 0-1	2 74.92930	.00000	3.35419	.007271	***	***	***	***	43.15, 61.92	61.92, 43.15
0 0 0 3 2 0 - 2 1, 2 0 - 2	2 43.01059	.00000	2.36950	•014570	***	***	***	***	61.92. 75.07	75.07.61.92
0 0 0 3 1 0-1-1, 1 0-1	4 36.81237	.00000	2,08141	.018882	***	***	1.7	1.7	25.11, 61.92	61.92 25.11
0 0 0 3 3-1-2-2, 3-1-2	5 23.26969	19.10661	1.37231	•043437	•80	•57	***	***	44.77, 68.04	68.04, 44.77
0 0 0 5 -1 0 1 4, -1 0 1	1 87.03429	.00000	2.08141	.018882	***	***	1.7	1.7	61.92, 25.11	25.11, 61.92
0 0 0 5 1 0-1 0, 1 0-1	5 69.44726	.00000	1,95154	.021479	***	***	***	***	20.55, 90.00	90.00, 20.55
0 0 0 5 2 0-2 2, 2 0-2	3 66.74139	.00000	1,91482	.022310	***	***	***	***	51.34, 61.92	61.92, 51.34
0 0 0 5 3-1-2 2, 3-1-2	3 53,12854	19.10661	1.66733	•029425	***	***	***	***	58.83, 68.04	68.04, 58.83
0 0 0 5 3-1-2 1, 3-1-2	4 50.28395	19,10661	1,60321	•031826	***	•79	***	***	51.12, 78.60	78.60, 51.12
0 0 0 5 4 0 - 4 2 + 4 0 - 4	3 36.73881	.00000	1.24670	•052630	***	.57	1.1	i.i	68.20. 75.07	75.07, 68.20
1 1-2 1 1 0-1-1	87.61075	33,17881	3.06457	.008710	***	***	1.6	***	38.67	53.72
1 1 - 2 1 1 0 - 1 - 2	64.05402 58.77035	21.42068	2.62289	•010753	***	***	***	***	46.51	69.43
1 1-2 1 1-1 0-1	52.35574	62.98760	2.42869	.013868	***	***	1.9	***	29.68	82.04
1 1-2 1 1 0-1-3	49.04656	15,65409	2.31651	.015244	***	• 35	1.4	***	51.85	79.11
1 1-2 1 1-1 0-2	43.71258	44.44408	2.11958	.018208	***	***	***	***	33.99	77.60
1 1-2 1 3-1-2 0 1 1-2 1 2 0-2-3	42.99673	26,12288	2.09172	+018696	.77	***	***	***	80.00 74.54	51.27
1 1-2 1 3-1-2-1	41.68316	62,98760	2.03974	.019661	***	•51	***	***	82.04	56.28
1 1-2 1 3-1-2-2	38.64252	52,59496	1,91536	•022298	***	***	1.9	1.8	79.40	61.95
	36.56302	21,42068	1.82717	•024502	.27	***	1.5	***	74.00	69.43
1 1-2 1 1 0-1-5	32.71264	10,11092	1.65761	+020515	.41	. 39	1.3	***	58.15	89.14
1 1-2 1 4-3-1 0	28.94434	85.83460	1.48442	.037124	- 26	***	1.4	***	74.33	76.73
1 1-2 1 2 1-3-6	25.80835	7.45033	1.33536	•045874	•31	•88	1.4	1.7	85.97	68.22
1 1-2 3 2 0-2 1	81.81508	51,91783	2.33327	.015026	***	***	***	***	60.32	37.86
1 1=2 3 2 0=2 2 1 1=2 3 1 0=1=1	76.17517	27.04277	2.28900	•015147	***	***	1.2	***	27.82	20.20
1 1-2 3 2 0-2 0, 0 1-1	3 74.63507	40,39036	2,27303	.015833	***	***	***	***	54.87, 50.50	50.50, 24.14
1 1-2 3 2 1-3 0	59.95112	15,83292	2,04046	.019647	***	•69	***	***	76.21	43.84
1 1 - 2 3 3 - 1 - 2 1	55.88924	68,60481	1.94801	+021475	***	***	***	***	71.37	52.74
1 1=2 3 3=1=2=1	49.38578	51.91783	1.78944	.025546	.78	***	***	***	60.32	70.29
1 1-2 3 3-2-1 0	47.75503	76.77114	1.74504	.026863	***	***	***	1.3	56.16	76.09
1 1-2 3 2 0-2-3	44.53988	23,04434	1.65341	•029923	***	***	1.9	1.1	49.63	85.83
1 1-2 3 4 0-4-3	52.50028	27,00120	1.20700	•051556	• 50	• 90	1.,	1.2	//.47	/0.21
2 2=4 1 0 2=2 3 2 2=4 1 2 0=2=3	87.11003	37.18783	1,58399	•032603	• 39	***	1.2	***	46.88	40.23
2 2-4 1 2 0-2-4	67.75458	22.85560	1.46796	.037961	•46	.89	1.7	1.8	50,56	61.69
2 2-4 1 3 1-4-3	61.64625	25,36991	1.39574	•041991	•86	***	1.3	1.7	82.08	36.27
2 2-4 1 2 1-3-5	58.74320	9.36583	1.35580	•044501	•41	.85	1.3	1.8	66.39	54.87
2 2-4 1 2 0-2-5	50.70058	19,02780	1.00010	1044542	• 52	***	1.5	1.0	54.30	00.94
3 0-3 1 1 0-1 3	86.07581	.00000	1,82006	•024694	•56	***	1.3	1.3	38.16	47.92
3 0-3 1 2 0-2 3	76.44624	.00000	1.77353	•026007	***	+20 ***	***	1 • / ***	50.95 54.A1	48.75
3 0-3 1 3-1-2-3	66.42141	23,27582	1.67202	.029260	.69	***	1.4	1.4	68.44	45.14
3 0-3 1 2 0-2-4	63.70010	.00000	1.63549	•030582	***	•89	***	***	59.36	56.93
3 0-3 1 3-2-1-3	60.7787 <u>1</u> 57.57757	41.96954	1.5921/	•032269	***	***	***	1.2	62.33	56.89
3 0-3 1 1 0-1-7	40.10352	.00000	1,17518	.059231	.62	.65	1.9	1.9	54.81	85.09
3 0-3 1 5-2-3 0	39.11927	67.98696	1.15104	.061742	. 39	.43	1.3	1.2	64.50	25.38
3 0-3 3' 3-2-1 1	85.96109	74.79800	1.63083	.030757	•30	***	***	1.1	52.36	41.68
3 0-3 3 3-2-1 0	81.51443	61,47756	1.61699	•031286	***	***	***	***	50.32	48.17
3 0-3 3 3-1-2-2	73.22353	22,23968	1.56531	+031468	.03	•90	1.3	1.4	17017	53.33
3 0-3 3 2-2 0-1	72.30721	61.47756	1.55756	•033719	***	***	1.5	1.3	35.45	72.24
3 0-3 3 0 1-1 1	70.69144	74.79800	1,54293	.034361	***	***	1.5	1.4	18.34	52.36
3 0-3 3 2 0-2-3	68.68643	.00000	1,52307	•035263	***	***	1.8	1.8	44.57	66.74
3 0-3 3 1 0-1-4	53.28534	04.02443	1.31057	.040062	•40	. 39	1.2	∓ +∓ 1.2	33.75	43.93 87.03
3 0-3 3 2 0-2-5	50.05343	.00000	1,25338	.052071	***	.42	1.0	1.0	48.73	81.21
3 0-3 3 4-1-3-4	49.73545	13.78469	1,24753	.052560	•51	.29	1.5	1.4	70.09	60.17

to a case of only 3 beams. However, all of the instances of intermediate back-reflections in Tables 3, 4, and 5 occur for coincident operating points corresponding to cases of 4 beams. Thus the argument must be modified to show that all non-vanishing 'routes' of multiple reflection going from the incident to the final beam include at least one back-reflection. For this purpose it is convenient to consider the first operating point in Table 3 as an example. There are four independent indirect routes, which may be identified with the following sequences of reflections:

$(10\overline{1}0) + (\overline{1}011)$	(3 <i>a</i>)
$(10\overline{1}1) + (\overline{1}010)$	(3 <i>b</i>)
$(10\overline{1}0) + (0001) + (\overline{1}010)$	(3 <i>c</i>)

 $(10\overline{1}1) + (000\overline{1}) + (\overline{1}011)$. (3*d*)

The first and second sequences correspond to the simulations given in Table 3; both include back-reflections and thus give good wavelength resolution. The third and fourth sequences are not high-resolution simulations. Fortunately, however, they contribute nothing, since the middle links in both sequences have vanishing structure factors. Thus high-resolution monochromat-

Table 5. Operating points free of interference for 8 simulated zinc reflections

See caption of Table 3.

REFLECTION	INDICES (HKI	L)	THETA			ENERGY		THETA+	BETA-	BETA+	THETA2	THETA3
PRIMART	SECONDARY		(DEG)	(DEG)	(4)	(27)	(DEG)	(DEG)	(020)	(0207	(020)	(000)
0 0 0 1 1	0-1 0, 1 0	-1 1	25.01085	.00000	4.18291	.004675	***	***	***	***	64.99, 90.00	90.00, 64.99
	0 = 1 = 1 + 1 + 1 = 0 0 = 2 = 1 + 2 = 0	-1 2	11.88032	.00000	2.03678	.019719	. 44	***	***	***	64.99. 76.87	76.87 64.99
	0=2=1, 20 0=1=2, 10	-22	11.43766	.00000	1.96192	.021252	***	.44	***	***	35.55. 46.98	46.98. 35.55
0 0 0 1 2	0-2-2, 20	-2 3	9.97376	.00000	1.71354	.027859	•51	***	***	***	55.02, 64.99	64.99, 55.02
0 0 0 3 1	0-1 1, 1 0	-1 2	68.02804	.00000	3,05833	.008746	***	***	***	***	46.98, 64.99	64.99: 46.98
0 0 0 3 1	0-10, 10	-13	54.45486	.00000	2.68333	.011361	***	***	***	***	35.55, 90.00	90.00, 35.55
0003 2	0-21, 20	-2 2	38.14138	.00000	2.03678	•019719	***	***	***	***	64.99, 76.87	76.87. 64.99
0003 1	0-1-1, 10	-1 4	20.88666	.00000	1.64328	.020957	***	***	***	***	46.98. 76.87	76.87 46.98
0 0 0 3 2	0-2-2. 20	-2 5	24.38006	.00000	1.36132	.044141	.41	***	***	***	40.61, 64.99	64.99, 40.61
0 0 0 3 4	0-4-1, 4 0	-4 4	18.35824	.00000	1,03869	.075822	•47	.93	1.6	1.6	64.99, 83.35	83.35, 64.99
0 0 0 5 1	0-1 1, 1 0	-1 4	86.82569	.00000	1,97568	.020957	***	***	1.8	1.8	28.19, 64.99	64.99, 28.19
0 0 0 5 1	0-1 0, 1 0	-1 5	66.79569	.00000	1,81865	•024732	***	***	***	***	23.20, 90.00	90.00, 23.20
0005 2	0-22, 20	-2 3	59,99545	.00000	1.71354	•027859	***	***	***	***	55.02, 64.99	64.99, 55.02
0005 2	0-21, 20	-2 4	56.14773	.00000	1,04328	•030293	***	***	1.7	***	40.901 /0.07	70.57. 62.12
0005 3	-1-2 2 = 3-1	-2 3	47.30024	19,10661	1,40721	.041309	***	.35	1.4	1.4	19.66. 64.99	64.99, 19.66
0005 1	-1-2-1. 3-1	-2 6	36.61413	19,10661	1.18015	.058733	***	.79	1.5	***	43.39, 80.00	80.00, 43.39
0 0 0 5 4	-1-3 2, 4-1	-33	35.72457	13,89788	1.15535	.061282	•68	.75	1.5	1.1	68.78, 75.49	75.49, 68.78
0 0 0 5 4	0-4 2, 4 0	-4 3	32.41566	.00000	1.06071	•072706	•29	***	***	***	70.71, 76.87	76.87, 70.71
0005 4	0-4 1+ 4 0	-4 4	31.66346	.00000	1.03869	•075822	***	.47	1.6	1.6	64.99, 83.35	83.35, 64.99
0 0 0 5 4	0-4→1, 4 0	-4 6	28.33200	.00000	• • 3 9 0 0	•092763	•55	• 99	1.7	1.9	55.021 85.55	03.337 33.02
1 1-2 1 0	1-1 2		87.54361	36,49949	2.57082	•012377 •013997	***	***	1.8	***	49.62	37.92
1 1=2 1 2	0 = 1 = 2	-1 1	63.05635	65.74927	2.29388	.015546	***	.88	***	1.1	83.69, 33.26	33.26, 29.80
1 1-2 1 1	0-1-3		54.55733	17.59488	2,09637	.018614	***	.87	1.8	***	51.38	74.07
1 1-2 1 1	-1 0-1		53.88729	65.74927	2.07877	•018930	***	***	***	1.5	29.80	83.69
1 1-2 1 1	-1 0-2		46.31672	47.98226	1,86085	•023623	***	***	***	***	33.46	79,78
1 1-2 1 2	2 0-2-3		45.66443	29,02855	1.79351	+024149	.94	***	***	***	55.35	57.57 80.46
1 1 - 2 1 1	2 0-1-4		39.54987	23.93971	1.63848	.030471	. 37	***	***	***	76.17	64.28
1 1-2 1 2	1-3-3		36.22874	13.85540	1,52078	.035370	***	.32	***	1+1	80.46	44,23
1 1-2 1 5	5-3-2 0		24.28874	83.57439	1.05844	.073018	•27	•63	***	***	88.27	67.44
1 1-2 3 1	0-1-1		80.87191	28,86169	2,04651	.019532	***	***	***	***	29.29	69.84
1 1-2 3 2	2 0-2 1		78.40239	54.02991	2.03044	•019842	***	***	***	***	64.61	36,99
1 1-2 3 2	2 0-2 2		76.59262	70.05555	2.01626	•020122	***	***	***	1.6	74.59 59.24. H7.66	20.82
1 1-2 3 2	2 0-2 0, 0 1	-1 3	73.10180	42.57045	1.87113	•020797	***	***	***	***	56.36	59.12
1 1=2 3 1	0-1-2		61.11913	21.48876	1.81496	.024833	***	***	***	***	32.53	86.35
1 1-2 3 3	-1-2 2		51,77868	79,71633	1.62841	.030849	.47	.87	***	***	81.79	46.43
11-23 3	5-1-2 0		51.73225	61,44015	1.62737	.030888	• 35	***	2.0	***	68.88	59.39
1 1-2 3 2	2 1-3-2		48.94388	11,96859	1,56300	•033485	***	***	1.9	***	71.80	59.25
1 1-2 3 1	1 0-1-3		40.00209	10 4102432	1.43555	.039694	***	.81	***	***	68.58	67.59
1 1-2 3 4	+-1-3-2		35.63606	46.66613	1,20766	.056089	. 39	.97	1.2	2.0	77.02	67.35
2 2=4 1 2	2 0-2-3		84.74224	31.71368	1,31497	.047307	•49	***	1.8	***	44.06	51,20
2 2-4 1 0	2-2 3		82.59803	40.86372	1.30952	.047702	***	•58	1.8	1.4	43.83	38.77
2 2-4 1 3	5-2-1-1		63.68664	76,98293	1.18370	•058382	•83	•42	1.1	1+1	43.55	72.77
2 2-4 1 1	0-1-6		55.13579	9,81635	1,08351	•069679	***	.50	1.0	1.8	44.25	80.62
3 0-3 1 2	2 0-2-3		83.20055	.00000	1.50961	.035895	•67	***	***	***	52.98	43.82
3 0-3 1 1	0-1-3		82.52794	.00000	1,0740	.036000	***	•67	***	***	20.08	59.16
30-31	J 1-1 1 1 0-1 3		79.46570	.00000	1,49468	+036615	.95	***	1.5	1.5	33.85	45.62
3 0-3 1 2	2-2 0-1		75.57461	70,46146	1,47238	.037733	***	.38	1.0	***	40.93	63.50
3 0-3 1 2	2 0-2-4		70.18711	.00000	1,43031	.039985	***	***	1.6	1.6	57.96	51.86
3 0-3 1	1 0-1 2		56.14773	,00000	1,26258	.051315	***	•40	1.0	1.0	21.97	34.18
3 0-3 3 2	2 0-2-2		88.19345	.00000	1.39361	•042119	•60	***	***	***	41.78	50.02
30-33	1 0-1-2		87.59189	.00000	1,39307	•042152	***	•60	***	***	24.38	00.03 39.66
50-33	3-2-1 0		81.00281	50.55811	1.37749	+042207	.03	●90 *±*	***	1.7	52.15	46.76
30-33	2-2 0-1		73.66026	63.97857	1,33799	.045694	***	***	1.4	***	36.54	69.80
3 0-3 3	0 1-1 1		72.41454	76,28247	1,32915	•046304	***	•88	1.4	1.9	18.53	53.89
3 0-3 3	2-2 0-4		51.89765	39.32926	1.09719	•067951	***	***	1.1	1.5	40.56	87.54
30-33 4	4 0-4-3		51.05605	.00000	1 05770	•069559	• 36	•35	1.0	1.1	04.60 87.35	44.30
30-33	2 0-2-2 0 2 0-2-7		47.34522	000000	.96122	.088535	•5/	•DC ***	1.6	1.6	52.89	83.53
						,						

ization can be obtained even with a mosaic crystal. Analogous considerations are valid for 9 other operating points in Tables 3, 4 and 5.

References

- BACON, G. E. (1962). *Neutron Diffraction*, 2nd ed., Chap. III & IV. Oxford: Clarendon Press.
- DAWSON, B. & WILLIS, B. T. M. (1967). Proc. Roy. Soc. A 298, 307.
- GUINIER, A. (1952). X-ray Crystallographic Technology, p. 79. London: Hilger & Watts. International Tables for X-ray Crystallography (1952). Vol.

I, p. 55. Birmingham: Kynoch Press. KOTTWITZ, D. A. (1968a). Acta Cryst. A 24, 117.

KOTTWITZ, D. A. (1968b). Phys. Rev. 175, 1056.

PEARSON, W. B. (1967). Handbook of Lattice Spacings and

Structures of Metals, Vol. 2, Chap. II. Oxford: Pergamon Press.

Physics Today (1964). Vol. 17, no. 2, p. 48. New York: American Institute of Physics.

RENNINGER, M. (1937). Z. Phys. 106, 141.

Acta Cryst. (1969). A25, 464

A Method of Calculating Molecular Crystal Structures

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A novel method of calculating the molecular position in a lattice of known dimensions is presented. The repulsive lattice energy is approximated by a sum of quadratic nonbonded interatomic potential functions and the lattice energy sum is minimized by full-matrix least squares. The convergence range from arbitrary trial models is greater than the previously used steepest descent method using (exp-6) nonbonded potentials. Greatly increased speed of convergence is also obtained because of the inclusion of off-diagonal terms and the small number of repulsive interactions which are considered. The calculated packing models are sufficiently accurate to serve as a starting point for structure factor least-squares refinement based on diffraction data.

Introduction

A situation frequently encountered in the study of molecular crystals by diffraction methods is that the molecular structure is already known approximately, or at least, plausible predictions of the molecular structure can be made from expected bond distances and angles. In order to verify the model and to refine the molecular structure from the diffraction data it is necessary to locate the molecules in the unit cell of the crystal. In addition, the packing structure may be of considerable interest in itself even if the molecular structure is accurately known. For example, several different packing structures may be observed for the same molecular structure. Or, chemical and physical interactions between molecules may be closely related to their mode of packing in the crystal.

The most obvious procedure which can be used to obtain the crystal packing of molecules is to minimize the lattice energy, neglecting thermal effects, using the best available representation of the non-bonded energy* between the molecules, and assuming pairwise additivity (Williams, 1965*a*). The most important con-

tribution to the attractive energy in molecular crystals is the London dispersion energy, which has an inverse sixth power dependence on the interatomic separation. This term has the physical meaning of instantaneous dipole-dipole polarization. Terms involving quadrupoles or higher may safely be neglected as an initial approximation.

Several investigators (Kitaigorodskii, 1965; Craig, Mason, Pauling & Santry, 1965) have shown that the contribution to the lattice energy from electrostatic dipoles or quadrupoles is small for typical molecular crystals. Further, the rate of change of electrostatic energy effects with respect to molecular position is small.

The repulsive energy is of primary importance in determining the molecular position, provided the observed lattice constants are retained. The repulsive energy is due to overlap of filled electron shells and is a consequence of the required antisymmetry of the wave function for the system. The increase in repulsion energy with decreasing interatomic separation, d, may be fitted by a d^{-n} term, with n chosen for best fit, or to an exponential exp (-Cd) term, with C chosen for best fit.

The pairwise sum for the lattice energy based on the usual models for the nonbonded energy, such as the

^{*} For a general discussion of nonbonded energy see, for example, Hirschfelder, Curtiss & Bird (1954).